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INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Mondet; Jean Aulnay Sous	Bois			FR
Lion; Bertrand	Livry Gargan			FR

ASSIGNEE-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY	TYPE CODE
L'Oreal	Paris			FR	03

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U.S. PATENT DOCUMENTS

PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<input type="checkbox"/> <u>4552755</u>	November 1985	Randen	
<input type="checkbox"/> <u>5116924</u>	May 1992	Bung et al.	

OTHER PUBLICATIONS

T. Shohei et al., "Hair Cosmetic", Patent Abstract of Japan for JP 08 092046, (Abstract Only), 96 (8), Aug. 30, 1996.

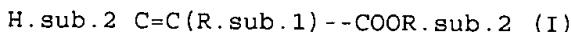
ART-UNIT: 167

PRIMARY-EXAMINER: Webman; Edward J.

ABSTRACT:

The subject of the present invention is the use, in, and for the preparation of, cosmetic or dermatological compositions, of a copolymer which can be obtained by copolymerization of a monomer mixture comprising:

- (a) from 5 to 25% by weight of at least one monomer (A) with ethylenic unsaturation containing at least one carboxylic acid function;
- (b) from 3 to 30% by weight of at least one monomer (B) of formula (I):



in which

$\text{R}.\sub{1}$ denotes a hydrogen atom or a methyl radical,

$\text{R}.\sub{2}$ denotes a saturated or unsaturated, linear or branched or cyclic alkyl radical or an aromatic radical, having from 8 to 40 carbon atoms,

- (c) at least one monomer (C) chosen from the group consisting of tert-butyl methacrylate, tert-butyl acrylate and mixtures thereof, in a proportion greater than 50%, such that the glass transition (T_g) of the polymer is greater than or equal to 15.degree. C.,

the weight percentages being calculated relative to the total amount of monomers used, as well as to the cosmetic or dermatological compositions used.

23 Claims, 0 Drawing figures

Exemplary Claim Number: 1

BRIEF SUMMARY:

- 1 The present invention relates to the use of acrylic copolymers in, and for the preparation of, cosmetic or dermatological compositions, as well as to the compositions used.
- 2 For many cosmetic applications, in particular those intended for the treatment and care of the hair, the skin or the eyelashes, polymers are used which are capable, after application to the support to be treated and drying, of forming a deposit having mechanical properties and adhesive properties. In this respect, polymers are sought which can be removed easily under the action of an aqueous solution of surfactants. In parallel, it is sought to obtain a film-forming deposit which is resistant to environmental moisture, in particular a deposit which does not feel hygroscopic, a rain-resistant deposit (for hair or skin use) or a deposit which is resistant to lachrymal fluid (mascaras). Moreover, a film-forming deposit is sought to provide cosmetic properties such as the soft feel generally imparted by hydrophobic substances in cosmetics.
- 3 Another problem relating to deposit polymers occurs in the field of hair care, in particular in the context of products for fixing the hair or for retaining its shape. The reason for this is that it is often difficult to adapt the properties of the polymer which needs to be deposited in order to obtain, at the same time, good fixing of the hair, good staying power of this fixing and easy removal by combing or brushing.

- 4 It is known to use acrylic copolymers in cosmetic compositions. By way of example, patent application JP-A-87-167 307 describes a styling polymer obtained by copolymerization of the monomer mixture comprising:
- 5 (i) from 10 to 50% by weight of unsaturated carboxylic acids,
- 6 (ii) from 10 to 70% by weight of acrylic acid esters and/or of methacrylic acid esters,
- 7 (iii) from 0 to 50% of vinyl monomers.
- 8 This polymer makes it possible to give the hair good slippery properties while at the same time maintaining a good feel. However, the presence of a fatty chain in the polymer, introduced by the acrylic acid esters and/or the methacrylic acid esters, makes it difficult to remove the polymer once fixed on the hair.
- 9 A cosmetic hair composition is also known from application JP-A-08092046, this composition containing a copolymer comprising from 6 to 35% by weight of (meth)acrylic acid or of itaconic acid, from 15 to 50% by weight of C_{sub}.10 -C_{sub}.18 -alkyl (meth)acrylate, from 15 to 50% by weight of C_{sub}.4 -C_{sub}.8 -alkyl (meth)acrylate and from 0 to 25% of vinyl-type monomer, in particular acrylamide-type monomer. However, these polymers are difficult to use in compositions comprising an alcoholic medium, but such compositions are particularly advantageous owing to their fast drying time.
- 10 One of the objects of the present invention is thus to use, in cosmetic or dermatological compositions, acrylic polymers having film-forming properties and adhesive and mechanical properties that are satisfactory, these polymers being easily removed under the action of an aqueous solution of surfactants, and being easily incorporated into compositions comprising at least one alcoholic medium. In addition, it is an object of the invention to use film-forming polymers which provide good cosmetic properties before and after removal of the composition.
- 11 Another object of the invention is to use, in and for the preparation of hairstyling compositions, polymers having both high-performance fixing properties which offer good resistance to weak mechanical stresses and good rigidity, but are then readily removed by shampooing, brushing or combing, while at the same time providing good cosmetic properties such as a soft feel. In particular, one of the objects of the invention is to propose the use of polymers which are easy to use in hair compositions comprising an alcoholic medium in order to allow rapid drying of the composition, the said composition having good lacquering power and no sensation of stickiness.
- 12 The Applicant has discovered, surprisingly, that these objects can be achieved by using certain particular acrylic copolymers in, and for the preparation of, cosmetic or dermatological compositions.
- 13 The subject of the present invention is the use, in, and for the preparation of, cosmetic or dermatological compositions, of a copolymer which can be obtained by copolymerization of a monomer mixture comprising:
- 14 (a) from 5 to 25% by weight of at least one monomer (A) with ethylenic unsaturation containing at least one carboxylic acid function;
- 15 (b) from 3 to 30% by weight of at least one monomer (B) of formula (I):
- H_{sub}.2 C=C(R_{sub}.1)--COOR_{sub}.2 (I)
- 16 in which
- 17 R_{sub}.1 denotes a hydrogen atom or a methyl radical,
- 18 R_{sub}.2 denotes a saturated or unsaturated, linear or branched or cyclic alkyl radical or an aromatic radical, having from 8 to 40 carbon atoms,

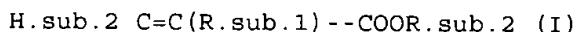
19 (c) at least one monomer (C) chosen from the group consisting of tert-butyl methacrylate, tert-butyl acrylate and mixtures thereof, in a proportion at least greater than 50% by weight, such that the glass transition (Tg) of the polymer is greater than or equal to 15.degree. C.,

20 the weight percentages being calculated relative to the total amount of monomers used.

21 The present invention also relates to cosmetic or dermatological compositions containing, in a cosmetically or dermatologically acceptable medium, at least one copolymer obtained by copolymerization:

22 (a) of 5 to 25% by weight of at least one monomer (A) with ethylenic unsaturation containing at least one carboxylic acid function;

23 (b) of 3 to 30% by weight of at least one monomer (B). of formula (I):



24 in which

25 R.₁ denotes a hydrogen atom or a methyl radical,

26 R.₂ denotes a saturated or unsaturated, linear or branched or cyclic alkyl radical or an aromatic radical, having from 8 to 40 carbon atoms,

27 (c) of at least one monomer (C) chosen from the group consisting of tert-butyl methacrylate, tert-butyl acrylate and mixtures thereof, in a proportion at least greater than 50% by weight, such that the glass transition (Tg) of the polymer is greater than or equal to 15.degree. C.,

28 the weight percentages being calculated relative to the total amount of monomers used.

29 It has been observed that the copolymer used according to the invention for hair compositions allows very good cosmetic properties to be obtained, in particular a good feel and good removal, both by brushing and by shampooing. In particular, after the composition has been applied, the hair has a pleasant feel without giving a "cardboard" effect, that is to say without making the hair excessively rigid. Thus, the composition has good lacquering power and, after it has been applied to the hair, the film formed is hardly sticky at all. In addition, the copolymer according to the invention shows good compatibility with alcoholic media and can thus be used readily in a composition with a fast drying time.

30 Other objects will become apparent on reading the description and the examples which follow.

31 Preferably:

32 the content of monomer (A) ranges from 6 to 20% by weight, and better still from 6 to 15% by weight,

33 the content of monomer (B) ranges from 5 to 25% by weight, and better still from 13 to 22% by weight,

34 the content of monomer (C) ranges from 55 to 80% by weight, and better still from 60 to 70% by weight.

35 The monomers (A) used to prepare the polymers according to the invention are chosen, for example, from the group consisting of: monocarboxylic acids with ethylenic unsaturation, such as acrylic acid, methacrylic acid and crotonic acid; dicarboxylic acids with ethylenic unsaturation, such as maleic acid, fumaric acid and itaconic acid and their monoester or monoamide derivatives with a C.₁-C.₄-alkyl group; and allyloxyacetic acid. Acrylic acid,

methacrylic acid or a mixture thereof is more particularly used.

- 36 In the monomers (B), R.sub.2 is preferably a linear or branched saturated alkyl radical having from 8 to 40 carbon atoms, and preferably from 8 to 30 carbon atoms. It may be, in particular, a radical corresponding to the Guerbet alcohols corresponding to the formula --CH₂--CH(R₃) (R₄) in which R₃ and R₄, which may be identical or different, denote a linear or branched, saturated alkyl radical, the total number of carbon atoms in R₃ and R₄ ranging from 6 to 38, and preferably from 6 to 28.
- 37 Advantageously, R₂ is chosen from the group formed by lauryl, stearyl and 2-ethylhexyl radicals.
- 38 Preferably, the copolymers used according to the invention consist of a mixture of monomers (A), (B) and (C) as defined above.
- 39 The copolymers according to the invention preferably have an average molecular weight, measured at the peak height by steric exclusion chromatography, ranging from 5,000 to 2,000,000, and in particular from 20,000 to 1,000,000.
- 40 Advantageously, the polymer according to the invention has a glass transition temperature ranging from 40.degree. C. to 90.degree. C.
- 41 The copolymer according to the invention can be obtained by radical polymerization of the monomers as defined above.
- 42 The radical polymerization can be carried out in solution in a solvent which is common to all the monomers employed and to the polymer obtained, or in a mixture of common solvents. As solvent, mention may be made of ethyl acetate or ethanol. The reaction is generally carried out at a temperature ranging from 30.degree. C. to the boiling point of the solvent employed. The polymerization can also be carried out in a heterogeneous medium, in particular in suspension, by precipitation or by emulsion.
- 43 The radical polymerization can be initiated by a conventional organic initiator, for example azobis-N-butyronitrile, bis(2-ethylhexyl) peroxydicarbonate or tert-butyl 2-peroxyethylhexanoate.
- 44 When the polymerization is carried out in emulsion, a water-soluble thermal initiator can also be used, such as potassium persulphate, aqueous hydrogen peroxide solution or a water-soluble redox system, for example one of persulphate/metabisulphite type. Moreover, the emulsion is stabilized in a known manner with a surfactant or a mixture of surfactants, those most commonly used being sodium lauryl sulphate and alkyl ethoxysulphates.
- 45 The duration of the polymerization reaction can range in general from 4 hours to 18 hours.
- 46 Advantageously, the copolymer according to the invention can be partially or totally neutralized depending on the desired solubility of the polymer in the composition. The degree of neutralization can range in particular from 30% to 100%. The neutralization can be carried out using an organic or inorganic base.
- 47 As inorganic base, mention may be made of sodium hydroxide or potassium hydroxide. As organic base, mention may be made of an amino alcohol taken from the group consisting of 2-amino-2-methyl-1-propanol (AMP), triethanolamine, triisopropanolamine ((TIPA)), monoethanolamine, diethanolamine, tris[1-(2-hydroxy)propyl] amine, 2-amino-2-methyl-1,3-propanediol (AMPD) and 2-amino-2-hydroxymethyl-1,3-propanediol.
- 48 The compositions according to the invention can also optionally contain a plasticizer in order to enhance the mechanical properties, the cosmetic properties and the adhesion to keratin substances of the deposited film-forming acrylic polymer after application and drying. The presence of a plasticizer is not obligatory in order to adjust the lacquering power in the lacquer

formulations of the invention, in contrast with conventional lacquer formulations.

- 49 Of the plasticizers which can be used according to the invention, mention may be made of:
- 50 Carbitols from the company Union Carbide, namely
- 51 Carbitol or diethylene glycol ethyl ether, methyl
- 52 Carbitol or diethylene glycol-methyl-ether, butyl
- 53 Carbitol or diethylene glycol butyl ether or hexyl
- 54 Carbitol or diethylene glycol hexyl ether,
- 55 the Cellosolves from the company Union Carbide, namely Cellosolve or ethylene glycol ethyl ether, butyl Cellosolve or ethylene glycol butyl ether, and hexyl Cellosolve or ethylene glycol hexyl ether,
- 56 propylene glycol derivatives and in particular propylene glycol phenyl ether, propylene glycol diacetate, dipropylene glycol butyl ether, tripropylene glycol butyl ether and the Dowanols from the company Dow Chemical, namely Dowanol PM or propylene glycol methyl ether, Dowanol DPM or dipropylene glycol methyl ether and Dowanol TPM or tripropylene glycol methyl ether.
- 57 Mention may also be made of:
- 58 diethylene glycol_methyl_ether or Dowanol DM from the company Dow Chemical,
- 59 castor oil oxyethylenated with 40 mol of ethylene oxide, such as that sold by the company Rhone Poulenc under the name "Mulgofen LE-719",
- 60 benzyl alcohol,
- 61 triethyl citrate sold by the company Pfizer under the name 1,3-butylene glycol,
- 62 diethyl, dibutyl and diisopropyl phthalates and adipates, *and*
- 63 diethyl and dibutyl tartrates,
- 64 diethyl, dibutyl and bis(2-ethylhexyl) phosphates, and
- 65 glycerol esters such as glyceryl diacetate (diacetin) and glyceryl triacetate (triacetin).
- 66 The plasticizers are chosen more particularly from those which are hydrophilic or water-soluble.
- 67 The plasticizer is present in a proportion preferably ranging from 0 to 20% by weight relative to the weight of the film-forming polymer. This proportion varies according to the application envisaged.
- 68 The cosmetic and dermatological compositions according to the invention thus contain, in a cosmetically or dermatologically acceptable support, the polymers as described above, for applications as varied as those encountered, for example, in the field of haircare, make-up or skincare, or any other cosmetic field in which it is desirable or sought to use a film-forming substance.
- 69 The copolymers according to the invention can be used alone as film-forming agents or alternatively as additives to conventional film-forming agents in, and for the preparation of, cosmetic or dermatological compositions.
- 70 Among the applications preferably envisaged by the present invention, mention may be made more particularly of:

- 71 the field of hair products (hair beauty, care or washing), in which the compositions according to the invention can be in the form of aerosols, foam, shampoos, conditioners, styling or treating lotions or gels, or shaping, setting or fixing lacquers or lotions.
- 72 the field of make-up products, in particular products for making up the nails, the eyelashes or the lips, in which the compositions according to the invention can be in the form of nail varnish; mascaras or eyeliners; lipsticks.
- 73 the field of skincare products (antisun products, sera, masks, lotions, milks, creams).
- 74 The copolymer is present in the cosmetic or dermatological compositions of the invention at a concentration generally ranging from 0.1 to 50%, and more preferably from 1 to 30%, by weight relative to the total weight of the composition. It varies depending on the cosmetic or dermatological application envisaged.
- 75 In the case of hair compositions, the polymer concentration can range from 0.5 to 25%, and in particular from 1 to 20%, by weight relative to the total weight of the composition.
- 76 In the case of nail varnishes, this proportion generally ranges from 2 to 35% by weight, and when the copolymer of the invention is used alone as film-forming agent, the concentration is equal to or greater than 30% by weight relative to the total weight of the composition.
- 77 In the case of mascaras or eyeliners, the polymer concentration generally ranges from 1 to 30% by weight relative to the total weight of the composition.
- 78 In the case of skincare compositions, the polymer concentration ranges from 0.5 to 20% by weight relative to the total weight of the composition.
- 79 The cosmetically acceptable support for the compositions according to the invention preferably consists of water, or one or more cosmetically acceptable organic solvents, or a mixture of water and one or more cosmetically acceptable organic solvents.
- 80 Among these organic solvents, C.sub.1 -C.sub.4 lower alcohols such as ethanol are more particularly used.
- 81 The copolymers according to the invention are dissolved or dispersed in the support for the compositions of the invention.
- 82 The compositions can also, of course, contain various adjuvants intended to make them acceptable in a particular cosmetic application.
- 83 The compositions according to the invention can contain conventional cosmetic additives chosen from fatty substances such as mineral, plant, animal or synthetic oils, animal, fossil, plant, mineral or synthetic waxes, organic solvents, thickeners, softeners, antifoaming agents, moisturizers, wetting agents, treating agents (agents for combating hair loss, antidandruff agents, etc.), antiperspirants, basifying agents, UV-A or UV-B or broad-band sunscreens, dyes, pigments, fragrances, plasticizers, preserving agents, anionic, nonionic or amphoteric organic polymers which are compatible with the copolymers of the invention, and propellants when the compositions are in aerosol form.
- 84 Needless to say, a person skilled in the art will take care to select the optional additional compound(s) mentioned above such that the advantageous properties intrinsically associated with the compositions according to the invention are not, or are not substantially, adversely affected by the addition(s) envisaged.
- 85 The subject of the invention is also a process for the cosmetic treatment of

keratin substances such as the skin, the hair, the scalp, the eyelashes, the eyebrows, the nails or the lips, characterized in that it consists in applying a composition as defined above to these.

DETAILED DESCRIPTION:

1 The examples which follow serve to illustrate the invention without, however, being limiting in nature.

2 PREPARATION EXAMPLES 1 TO 5

3 Example 1

4 Preparation of a Copolymer Obtained from a Mixture Consisting of:

5 15% by weight of acrylic acid

6 65% by weight of tert-butyl acrylate

7 20% by weight of lauryl acrylate

8 15 g of acrylic acid, 65 g of tert-butyl acrylate and 20 g of lauryl acrylate were introduced successively into a reactor under a stream of nitrogen. 200 g of ethyl acetate and 2 ml of tert-butyl peroxy-2-ethylhexanoate, sold under the name "Trigonox 21 S" by the company Akzo, were then added. The mixture was stirred at room temperature in order to obtain a homogeneous medium. It was then heated to the reflux point of the ethyl acetate in order to carry out the polymerization for 12 hours. The reaction medium was then diluted by adding 100 g of ethyl acetate and was then cooled to room temperature.

9 The reaction solution was purified by precipitation from 8 litres of petroleum ether. The precipitated polymer obtained was dried in an oven. 91 g of polymer having the following characteristics were obtained:

10 Acid number: 137.5

11 Tg=44.degree. C.

12 The characterization of the molecular weight was carried out by steric exclusion chromatography on a microstyragel column eluted with tetrahydrofuran. The results are expressed relative to a polystyrene standard.

13 The polymer obtained gives a main peak corresponding to a molecular weight of 322,000 and a shoulder towards high molecular weights corresponding to a molecular weight of 1,308,000.

14 Example 2

15 Preparation of a Copolymer Obtained from a Mixture Consisting of:

16 15% by weight of acrylic acid

17 70% by weight of tert-butyl acrylate

18 15% by weight of stearyl methacrylate

19 The polymer was prepared according to the same operating conditions as in Example 1. 90 g of polymer having the following characteristics were obtained:

20 Acid number: 130

21 Tg=53.degree. C.

22 Molecular weight: main peak: MW=182,000

23 shoulder: 1,030,000

24 Example 3

25 Preparation of a Copolymer Obtained from a Mixture Consisting of:

26 15% by weight of acrylic acid

27 70% by weight of tert-butyl acrylate

28 15% by weight of stearyl acrylate

29 The polymer was prepared according to the same operating conditions as in Example 1. 90 g of polymer having the following characteristics were obtained:

30 Acid number: 129

31 Tg52.degree. C.

32 Molecular weight: main peak: MW=160,000; no shoulder

33 Example 4

34 Preparation of a Copolymer Obtained from a Mixture Consisting of:

35 15% by weight of acrylic acid

36 65% by weight of tert-butyl acrylate

37 20% by weight of lauryl acrylate

38 The polymer was prepared according to the same operating conditions as in Example 1, replacing the ethyl acetate with ethanol. 80 g of polymer having the following characteristics were obtained:

39 Acid number: 139.5

40 Tg=44.degree. C.

41 Molecular weight: main peak: MW=96,000; no shoulder

42 Example 5

43 Preparation of a Copolymer Obtained from a Mixture Consisting of:

44 20% by weight of acrylic acid

45 60% by weight of tert-butyl acrylate

46 20% by weight of 2-ethylhexyl acrylate

47 The polymer was prepared according to the same operating conditions as in Example 4, using only 1 ml of initiator. 80 g of polymer having the following characteristics were obtained:

48 Acid number: 161

49 Molecular weight: main peak: MW=113,000; no shoulder

50 COMPOSITION EXAMPLES 6 TO 12

51 Example 6

52 Aerosol Styling Lacquer

Composition A

Copolymer of Example 1	2	g
2-Amino-2-methyl-1-propanol for 100% neutralization	qs	
Ethanol	qs	100 g

- 53 An aerosol hair lacquer was prepared by packaging the following in a suitable aerosol container:

Composition A	65	g
Dimethyl ether	35	g

- 54 The valve was fixed on and the container sealed hermetically. When applied to the hair, the lacquer has good lacquering power and good staying power. The polymer film is easily removed by brushing. The hair feels attractive, even after brushing.

55 Example 7

56 Aerosol Styling Lacquer

Composition B

Copolymer of Example 2	2	g
2-Amino-2-methyl-1-propanol for 100% neutralization	qs	
Ethanol	qs	100 g

- 57 An aerosol hair lacquer was prepared by packaging the following in a suitable aerosol container:

Composition B	65	g
Dimethyl ether	35	g

- 58 The valve was fixed on and the container sealed hermetically. When applied to the hair, the lacquer has good lacquering power and good staying power. The polymer film is easily removed by brushing. The hair feels attractive, even after brushing.

59 Example 8

60 Aerosol Styling Lacquer

Composition C

Copolymer of Example 3	2	g
2-Amino-2-methyl-1-propanol for 100% neutralization	qs	
Ethanol	qs	100 g

- 61 An aerosol hair lacquer was prepared by packaging the following in a suitable aerosol container:

Composition C	65	g
Dimethyl ether		35 g

- 62 The valve was fixed on and the container sealed hermetically. When applied to the hair, the lacquer has good lacquering power and good staying power. The polymer film is easily removed by brushing. The hair feels attractive, even after brushing.

63 Example 9

64 Aerosol Styling Lacquer

Composition D

Copolymer of Example 5	8.6	g
2-Amino-2-methyl-1-propanol for 50% neutralization	qs	
Ethanol	qs	100 g

- 65 An aerosol hair lacquer was prepared by packaging the following in a suitable aerosol container:

Composition D	37	g
Dimethyl ether	43	g
Pentane	20	g

- 66 The valve was fixed on and the container sealed hermetically. When applied to the hair, the lacquer has good lacquering power and good staying power, without cosmetic defects such as powdering or stickiness. The polymer film is easily removed by brushing. The hair feels attractive, even after brushing.

67 Example 10

68 Aerosol Styling Lacquer

Composition E

Copolymer of Example 4	8.6	g
2-Amino-2-methyl-1-propanol for	qs	
50% neutralization		
Ethanol	qs	100 g

- 69 An aerosol hair lacquer was prepared by packaging the following in a suitable aerosol container:

Composition E	37	g
Dimethyl ether	43	g
Pentane	20	g

- 70 The valve is fixed on and the container sealed hermetically. When applied to the hair, the lacquer has good lacquering power and good staying power, without cosmetic defects such as powdering or stickiness. The polymer film is easily removed by brushing. The hair feels attractive, even after brushing.

71 Example 11

72 Mascara

73 1) Preparation of an Aqueous Dispersion of the Polymer of Example 4:

74 30 g of the polymer of Example 4 are added to a solution of 90 g of methyl ethyl ketone and 1.76 g of 2-amino-2-methyl-1-propanol (the amount corresponding to 30% neutralization according to the acid number). After stirring at room temperature for 30 minutes, the polymer has completely dissolved.

75 An aqueous phase is stirred into the organic phase thus obtained using an Ultra-Turrax type stirrer at 2000 rev/min in order to prepare the emulsion, this emulsion consisting of 120 g of deionized water. After addition of the aqueous phase is complete, at room temperature, stirring is continued for 10 to 15 min, which allows a translucent, stable emulsion to be obtained.

76 Concentration is then carried out using a rotary evaporator under partial vacuum at a temperature below 50.degree. C. After removal of the methyl ethyl ketone, a stable dispersion whose polymer concentration is 20% by weight relative to the total weight of the composition is obtained.

77 The particle size is measured by quasi-elastic light scattering using a model M4 Coulter, and gives the following results:

78 Particle size: 45 nm.

79 Size polydispersity: 0.26

80 2) Preparation of the Mascara:

Triethanolamine stearate	12 g
Beeswax	6 g
Carnauba wax	1 g
Paraffin	3.5 g
Part B	
Iron oxides	6 g
Part C	
Hydroxyethylcellulose ("Cellosize QP" from Amerchol)	1 g
Gum arabic	2 g
Keratin hydrolysate	1.8 g
Part D	
Aqueous dispersion of the polymer of Example 4	5 g
Preserving agents	qs
Water	qs 100 g

- 81 This mascara is obtained by heating the ingredients of Part A at 85.degree. C., to which is added Part B, and the mixture is stirred using a turbomixer. The water in the preparation is then boiled and the preserving agents added, followed, at 85.degree. C., by the ingredients of Part C. The aqueous phase obtained is then added to Part A with stirring using a turbomixer, after which the aqueous dispersion of polymer of Part D is added and the mixture is paddle-stirred.
- 82 Example 12
- 83 Nailcare
- 84 A care base for the nails having the following composition is prepared:

Aqueous dispersion of the polymer of 82.3 g	
Example 4 (prepared according to Example 11.1)	
Water	15 g
Glycerol	2 g
Formaldehyde	0.5 g
Hydroxypropylcellulose	0.2 g

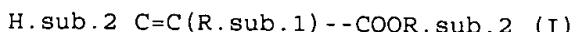
- 85 The aqueous dispersion, the glycerol and the formaldehyde are mixed together, followed by gentle stirring, and the hydroxypropylcellulose is then dispersed in the solution obtained.
- 86 This composition applies easily to the nails and allows a shiny film to be obtained which is easily removed with water. Daily application of this composition to the nails for several weeks hardens the nails.

CLAIMS:

We claim:

1. A cosmetic and/or dermatological composition comprising, in a cosmetically and/or dermatologically acceptable medium, at least one copolymer formed from the following monomer mixture:
 - (i) from 5 to 25% by weight of at least one monomer (A) with ethylenic unsaturation containing at least one carboxylic acid function;

(ii) from 3 to 30% by weight of at least one monomer (B) of formula (I):



in which:

R.₁ denotes a hydrogen atom or a methyl radical, and

R.₂ denotes a saturated or unsaturated, linear or branched or cyclic alkyl radical or an aromatic radical having from 8 to 40 carbon atoms; and

(c) more than at least 50% by weight of at least one monomer (C) selected from tert-butyl methacrylate and tert-butyl acrylate,

wherein the glass transition (T_g) of said at least one copolymer is at least 15.degree. C., and further wherein said percentages of monomers (A), (B), and (C) are calculated relative to the total amount of monomers used to form said at least one copolymer.

2. A cosmetic and/or dermatological composition according to claim 1, wherein said amount of said at least one monomer (A) ranges from 6 to 20% by weight, said at least one monomer (B) ranges from 5 to 25% by weight, and said at least one monomer (C) ranges from 55 to 80% by weight, relative to the total weight of monomers used to form said at least one copolymer.

3. A cosmetic and/or dermatological composition according to claim 2, wherein said at least one monomer (A) ranges from 6 to 15% by weight, said at least one monomer (B) ranges from 13 to 22% by weight, and said at least one monomer (C) ranges from 60 to 70% by weight, relative to the total weight of monomers used to form said at least one copolymer.

4. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one monomer (A) is selected from monocarboxylic acids with ethylenic unsaturation; dicarboxylic acids with ethylenic unsaturation and the monoester and monoamide derivatives thereof with a C.₁-C.₄ alkyl group; and allyloxyacetic acid.

5. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one monomer (A) is selected from acrylic acid and methacrylic acid.

6. A cosmetic and/or dermatological composition according to claim 1, wherein, in said at least one monomer (B), R.₂ is a linear or branched saturated alkyl radical having from 8 to 40 carbon atoms.

7. A cosmetic and/or dermatological composition according to claim 6, wherein R.₂ is a linear or branched saturated alkyl radical having from 8 to 30 carbon atoms.

8. A cosmetic and/or dermatological composition according to claim 6, wherein, in said at least one monomer (B), R.₂ is selected from lauryl, stearyl and 2-ethylhexyl radicals.

9. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one copolymer has an average molecular weight, measured at the peak height by steric exclusion chromatography, ranging from 5,000 to 2,000,000.

10. A cosmetic and/or dermatological composition according to claim 9, wherein said at least one copolymer has an average molecular weight, measured at the peak height by steric exclusion chromatography, ranging from 20,000 to 1,000,000.

11. A cosmetic and/or dermatological composition according to claim 1, wherein said glass transition (T_g) ranges from 40.degree. C. to 90.degree. C.

12. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one copolymer is present in an amount ranging from 0.1 to 50% by weight relative to the total weight of said composition.
13. A cosmetic and/or dermatological composition according to claim 12, wherein said at least one copolymer is present in an amount ranging from 1 to 30% by weight, relative to the total weight of said composition.
14. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one copolymer is neutralized to a degree ranging from 30 to 100%.
15. A cosmetic and/or dermatological composition according to claim 1, wherein said cosmetically and/or dermatologically acceptable medium comprises water, at least one cosmetically or dermatologically acceptable organic solvent, or a mixture thereof.
16. A cosmetic and/or dermatological composition according to claim 15, wherein said at least one cosmetically or dermatologically acceptable organic solvent is selected from C.sub.1 -C.sub.4 lower alcohols.
17. A cosmetic and/or dermatological composition according to claim 1, wherein said at least one copolymer is dissolved or dispersed in said cosmetically and/or dermatologically acceptable medium.
18. A cosmetic and/or dermatological composition according to claim 1, wherein said composition further comprises at least one plasticizer.
19. A cosmetic and/or dermatological composition according to claim 1, wherein said composition further comprises at least one conventional cosmetic additive selected from fatty substances; organic solvents; thickeners; softeners; antifoaming agents; moisturizers; wetting agents; treating agents; antiperspirants; basifying agents; UV-A and UV-B broad-band sunscreens; dyes; pigments; fragrances; preserving agents; anionic, nonionic and amphoteric organic polymers which are compatible with said at least one copolymer; and propellants.
20. A cosmetic and/or dermatological composition according to claim 19, wherein said fatty substances are selected from plant, animal, mineral and synthetic oils and animal, fossil, plant, mineral or synthetic waxes.
21. A cosmetic and/or dermatological composition according to claim 1, wherein said composition is a hair composition, a make-up composition, or a skin care composition.
22. A method of preparing a cosmetic and/or dermatological composition comprising including in said composition at least one copolymer according to claim 1 as a film-forming agent or as an additive of a film-forming agent.
23. A method for cosmetically treating a keratin substance comprising applying an effective amount of a composition according to claim 1 to said keratin substance.

the following formula:

$$n = w_1 n_1 + w_2 n_2 + w_3 n_3 + \dots$$

where the mass fractions $w_1 + w_2 + w_3 + \dots = 1$ and n_1, n_2 are the refractive indices of the homopolymers.

A mean particle size can also be determined from *static light scattering* (measurement range: 300 – 1500 nm). Hydrodynamic diameters can be accurately measured by *quasielastic light scattering* (QELS); measurement range: 10 – 3000 nm) [84][85][86][87][88]

The particle size distribution can be determined with the *ultracentrifuge* (measuring range: 10 – 20 000 nm in a gravity field of 20 – 200 000 g) [89]. The particles must be spherical.

The surface areas of latex (dispersion) particles are determined by *soap titration*, from which the particle size can be calculated [90][91][92][93]. Soap titration is simple and quick to perform, and yields reasonable values with homologous polymer dispersion series. Care should be taken, however, when evaluating latex surfaces of different hydrophilicity.

Viscosity. Dispersions with solids contents of 40 – 60 wt % exhibit a broad spectrum of rheological behavior [94], [95] that depends on the volume fraction, temperature, particle size, particle size distribution, and auxiliary system. Dispersions may be highly liquid to pasty, and may exhibit a Newtonian to viscoelastic flow behavior [96]. In dispersions structural viscosity ("shear thinning") is frequently observed; dilatancy is observed less often; thixotropy and rheopexy are rarely encountered.

Stability. Dispersions must have a certain degree of mechanical (shear) stability if coagulation is not to occur during pumping, stirring, or spraying. Mechanical stability is checked by determining the amount of coagulate after a stirring test; the stirrer, temperature, shear velocity, time, and other parameters should correspond to possible stresses occurring during use.

Thermal stability is checked by subjecting the dispersions to the anticipated service temperature conditions. A freeze – thaw test with five cycles is widely used. However, the results depend on the container size and temperature control so that the correlation between laboratory results and frost stability under relevant weathering conditions is generally unsatisfactory.

The polymer dispersion must be stable to electrolytes when diluted with hard water and when formulating products such as paints or adhesives. The stability is therefore tested by adding salts.

Minimum Film-Forming Temperature. Whether a dispersion forms a coherent film after evaporation of the water or whether a brittle, pulverizable layer is formed depends on the glass transition temperature of the polymer, residual water content, and temperature at which the film is formed. The lowest temperature at which a coherent film can still form from the copolymer dispersion is called the minimum film-forming temperature (MFT). The MFT is measured with a metal plate that has a temperature gradient [97], [98]. The temperature at which the film begins to crack or the so-called white point, at which the turbid film starts to become clear, is measured [99]. The MFT depends on the copolymer composition, as well as the particle size and polarity of the comonomers [100][101][102]. It is usually comparable to the glass transition temperature, but may be substantially lower, probably because the emulsifier and water act as a plasticizer.

⇒ Continued....

[Return to Article](#)

[79] : J. Hansmann, *Adhäsion* **21** (1977) 272.

[Return to Article](#)

[80] : G. Stoev, M. Angelova, *HRC CC J. High Resolut. Chromatogr. Chromatogr. Commun.* **10** (1987) 25.

[Return to Article](#)

[81] : T. R. Crompton: *Analysis of Polymers*, Pergamon Press, Oxford 1989, chap. 9.

[Return to Article](#)

[82] : E. B. Bradford, J. W. Vanderhoff, *J. Polym. Sci. Polym. Symp.* **3** (1963) 41.

[Return to Article](#)

[83] : M. Kerker: "The Scattering of Light and other Electromagnetic Radiation," in E. M. Loebl (ed.): *Physical Chemistry*, Academic Press, New York 1969.

[Return to Article](#)

[84] : R. Pecora (ed.): *Dynamic Light Scattering: Applications of Photon Correlation Spectroscopy*, Plenum Press, New York 1985.

[Return to Article](#)

[85] : R. S. Stock, W. H. Ray, *J. Polym. Sci. Polym. Phys. Ed.* **23** (1985) 1393.

[Return to Article](#)

[86] : P. G. Cummins, E. J. Staples, *Langmuir* **3** (1987) 1109.

[Return to Article](#)

[87] : S. E. Bott, *Polym. Mater. Sci. Eng.* **53** (1985) 68.

[Return to Article](#)

[88] : J. Wagner, *Chem. Ing. Tech.* **58** (1986) 578.

[Return to Article](#)

[89] : W. Mächtle, *Makromol. Chem.* **185** (1984) 1025.

[Return to Article](#)

[90] : T. R. Paxton, *J. Colloid Interface Sci.* **31** (1969) 19.

BEST AVAILABLE COPY

[Return to Article](#)

[91] : J. Kloubek, K. Friml, S. Petrikova, F. Krejci, *J. Polym. Sci. Polym. Phys. Ed.* **14** (1976) 1451.

[Return to Article](#)

[92] : J. G. Brodnyan, G. L. Brown, *J. Colloid Sci.* **15** (1960) 76.

[Return to Article](#)

[93] : H. Schuller, *Kolloid Z. Z. Polym.* **211** (1966) 113.

[Return to Article](#)

[94] : I. M. Krieger, *Adv. Colloid Interface Sci.* **3** (1972) 111.

[Return to Article](#)

[95] : J. Mewis, A. J. B. Spaull, *Adv. Colloid Interface Sci.* **6** (1976) 173.

[Return to Article](#)

[96] : F. B. Malihi, T. Provder, M. E. Koehler, *J. Coat. Technol.* **55** (1983) no. 702, 41.

[Return to Article](#)

[97] : T. F. Protzman, G. L. Brown, *J. Appl. Polym. Sci.* **4** (1960) 81.

[Return to Article](#)

[98] : G. Taschen, *Plaste Kautsch.* **24** (1977) 212.

[Return to Article](#)

[99] : DIN 53 787, 1974.

[Return to Article](#)

[100] : J. G. Brodnyan, T. Konen, *J. Appl. Polym. Sci.* **8** (1964) 687.

[Return to Article](#)

[101] : S. Eckersley, A. Rudin, *Polym. Mater. Sci. Eng.* **58** (1988) 1115.

[Return to Article](#)

[102] : DIN 53 765, 1991.

[Return to Article](#)

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POLYACRYLATES - Properties (Erich Penzel)

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4. Properties

4.1. Polymer Solutions and Dispersions

4.1.1. Polymer Solutions

The flow properties of polymer solutions and their formulations are of great importance as regards processability. The temperature dependence of these properties determines the temperature range in which the solution polymer can be handled.

The viscosity of polymer solutions depends on the molecular mass, molecular structure, temperature, and thermodynamic behavior of the solvent [72]. Information on polymer molecular mass, coil dimensions, interaction between solvents and polymers, and the thermodynamic behavior of a solvent are usually obtained from measurements on dilute solutions (light scattering, osmometry, sedimentation, viscosimetry), see also → Plastics, Analysis.

Interactions between the polymer molecules increase with rising concentration. Temporary polymer – polymer contacts are formed that finally lead to the formation of undesirable gels that do not flow and cannot be processed.

4.1.2. Polymer Dispersions

Dispersions consist of two phases, and are therefore more complicated systems than polymer solutions; they often also contain polyelectrolytes. Almost all investigations are carried out on dilute systems.

The most important properties of polymer dispersions are appearance, concentration, pH, surface tension [79], residual monomer content, particle size, particle size distribution, viscosity, and stability.

Residual Monomer Content. Residual monomers are determined by chromatography. The volatile monomers can be quantitatively determined by gas chromatography using calibration and addition of an internal standard (e.g., toluene or methyl ethyl ketone [80]). An accuracy of ca. 1 ppm can be achieved.

The content of water-soluble, nonvolatile monomers (e.g., acrylic acid) is determined by high-pressure liquid chromatography [81].

Particle Size and Particle Size Distribution. The particle size and particle size distribution of emulsion polymers influence other properties (e.g., the viscosity of dispersions with a high solids content). At the same concentration and with the same auxiliary system, coarse dispersions generally have a lower viscosity than fine dispersions due to electrostatic effects or, in the case of dispersions with protective colloids, to steric effects.

Particle size and particle size distribution can be determined by *electron microscopy* (measurement range: 1 – 1000 nm) [82].

A simple method for determining the mean particle size is to measure the *light transmission* of a highly dilute sample. The relationship between the turbidity and particle radius of spheres is described by Mie's theory [83]. The particle diameter is shown as a function of the light transmission in Figure (2). An important parameter is m , the ratio of the refractive index of the polymer to that of water. The approximate refractive index of copolymers can be calculated with

WEST**End of Result Set**

L5: Entry 1 of 1

File: DWPI

May 15, 2002

DERWENT-ACC-NO: 2001-291904

DERWENT-WEEK: 200239

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TITLE: Aqueous cosmetic composition, e.g. nail varnish or hair setting lotion, contains an emulsion polymer with a glass transition point above room temperature and a minimum film-forming temperature below this

INVENTOR: SANNER, A; SCHLARB, B; WITTELER, H

PATENT-ASSIGNEE:

ASSIGNEE	CODE
BASF AG	BADI

PRIORITY-DATA: 1999DE-1039326 (August 19, 1999)

PATENT-FAMILY:

PUB-NO	PUB-DATE	LANGUAGE	PAGES	MAIN-IPC
EP 1204400 A2	May 15, 2002	G	000	A61K007/043
DE 19939326 A1	February 22, 2001		018	A61K007/043
WO 200113863 A2	March 1, 2001	G	000	A61K007/00

DESIGNATED-STATES: AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU MC NL PT RO SE SI CN
JP RO SI US AT BE CH CY DE DK ES FI FR GB GR IE IT LU MC NL PT SE

APPLICATION-DATA:

PUB-NO	APPL-DATE	APPL-NO	DESCRIPTOR
EP 1204400A2	August 18, 2000	2000EP-0965885	
EP 1204400A2	August 18, 2000	2000WO-EP08094	
EP 1204400A2		WO 200113863	Based on
DE 19939326A1	August 19, 1999	1999DE-1039326	
WO 200113863A2	August 18, 2000	2000WO-EP08094	

INT-CL (IPC): A61 K 7/00; A61 K 7/043; A61 K 7/11; A61 K 9/107

ABSTRACTED-PUB-NO: DE 19939326A

BASIC-ABSTRACT:

NOVELTY - Aqueous cosmetic compositions containing an emulsion polymer (A) with a minimum film-forming temperature (MFT) determined in the absence of auxiliary film-forming substances and at least one glass transition point (Tg) for the dried film, in which Tg is between 35 deg. C and 80 deg. C and the MFT is at least 8 deg. C below Tg.

USE - Emulsion polymers as described are used as film-forming agents in cosmetic compositions, especially aqueous compositions (claimed). Preferred applications are in nail varnish or hair setting formulations, e.g. hair sprays, lotions, setting foam and gels.

subject matter relied upon in the rejection. See MPEP § 201.11, 901.04 and 2127.

2. If the potential § 102(e) prior art reference is a **WIPO publication** of an International (PCT) Application under PCT Article 21(2) (e.g. has a “WO” publication number) with an international filing date **prior to November 29, 2000**, this publication may **not** be applied as prior art under § 102(e). Examiners can apply such a publication under § 102(a) or (b) as of its publication date. Alternatively, examiners can obtain a patent family search to determine if there is an earlier disclosure of the same subject matter. The Electronic Information Center (EIC) in each TC can perform patent family searches for examiners.
3. If the potential § 102(e) prior art reference is one of the following:
 - a. a **U.S. patent** that issued from an International (PCT) Application or claimed benefit to an International (PCT) Application;
 - b. a **publication of a U.S. application** under 35 U.S.C. § 122(b), including the publication of the National Stage (§ 371) of an International (PCT) Application; or
 - c. a **WIPO publication of an International (PCT) Application** under PCT Article 21(2) (e.g. has a “WO” publication number) with an **international filing date on or after November 29, 2000**,

examiners should:

- a. First, see if there is another reference that qualifies as prior art under § 102(a) or (b) that can be applied to the claims by:
 1. Obtaining a patent family search to determine if there is an earlier disclosure of the same subject matter that qualifies as prior art under § 102(a) or (b); or
 2. Finding equivalent prior art reference(s) that qualifies as prior art under § 102(a) or (b)
- b. If an equivalent § 102(a) or (b) reference cannot be located, the examiner should see the § 102(e) expert in their TC to determine if the potential § 102(e) prior art reference can be applied in a rejection.

When the final language of § 102(e) is resolved, all examiners will receive training on this subject matter, including training materials with detailed examples.

WEST Search History

DATE: Monday, September 23, 2002

<u>Set Name</u>	<u>Query</u>	<u>Hit Count</u>	<u>Set Name</u>
			result set
side by side			
	<i>DB=USPT; PLUR=YES; OP=ADJ</i>		
L23	6267950	1	L23
	<i>DB=DWPI; PLUR=YES; OP=ADJ</i>		
L22	schlarb.in. and sanner.in.	1	L22
	<i>DB=PGPB; PLUR=YES; OP=ADJ</i>		
L21	sanner.in.	11	L21
L20	schlarb.in.	2	L20
	<i>DB=USPT; PLUR=YES; OP=ADJ</i>		
L19	sanner.in. and keratin not L17	3	L19
L18	L16 and keratin not L17	0	L18
L17	L16 and nail	1	L17
L16	schlarb.in.	21	L16
	<i>DB=USPT,PGPB; PLUR=YES; OP=ADJ</i>		
L15	L14	1	L15
L14	L13 and nail	1	L14
L13	schlarb.in.	23	L13
L12	L11 and schlarb.in.	0	L12
L11	sanner.in.	181	L11
	<i>DB=USPT; PLUR=YES; OP=ADJ</i>		
L10	(L2 and L3 and L4 and L5 and L6 and L7).ab,clm.	3	L10
L9	L2 same L3 same L4 same L5 same L6 same L7	168	L9
L8	L2 and L3 and L4 and L5 and L6 and L7	255	L8
L7	styrene	121572	L7
L6	tbutylacrylate or (t adj butylacrylate) or (t adj butyl adj acrylate) or (tbutyl adj acrylate)	1963	L6
L5	methylmethacrylate or (methyl adj methacrylate)	48470	L5
L4	lauryl adj acrylate	3143	L4
L3	(nbutyl adj acrylate) or nbutylacrylate or (n adj butyl adj acrylate)	8627	L3
L2	acrylic	148640	L2
L1	6203720[pn]	1	L1

END OF SEARCH HISTORY